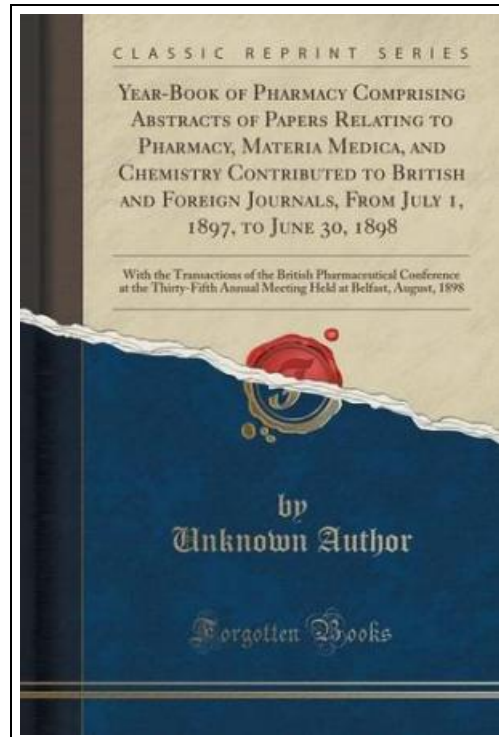


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Forgotten Books, United States, 2015. Paperback. Condition: New. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*. Excerpt from Year-Book of Pharmacy Comprising Abstracts of Papers Relating to Pharmacy, Materia Medica, and Chemistry Contributed to British and Foreign Journals, From July 1, 1897, to June 30, 1898: With the Transactions of the British Pharmaceutical Conference at the Thirty-Fifth Annual Meeting Held at Belfast, August, 1898 The latter, owing to their instability, are soon decomposed, yielding iodides and iodates, a change which is greatly accelerated by heat. Percarbonates, a new class of oxidising substances, are reported upon by M. Constam and A. von Haussen, who have obtained these bodies in the electrolysis of saturated solutions of alkaline carbonates at a low temperature. It splits up, when heated, into potassium carbonate, carbonic anhydride, and oxygen, and decomposes water at the ordinary temperature with liberation of oxygen and the formation of potassium bicarbonate. On account of this property it is recommended as a delicate reagent for the detection of moisture, especially in such substances as ether or chloroform. The action of magnesium on cupric sulphate solution, which results in the evolution of hydrogen and the precipitation of a mixture of cuprous oxide and metallic copper, has been investigated by F. Clowes and R. M. Caven, and is also discussed in its theoretical aspects by E. Divers. M. Francois directs attention to the fact that pure mercurous iodide has a bright yellow colour resembling that of lead chromate, and that the greenish colour of the commercial salt is due to contamination with metallic mercury. A very satisfactory product, quite free from this impurity and also from mercuric iodide, may be readily prepared by precipitation. The frequent occurrence of alkalis, sulphides, arsenic and other impurities in commercial samples of reduced iron is pointed...

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