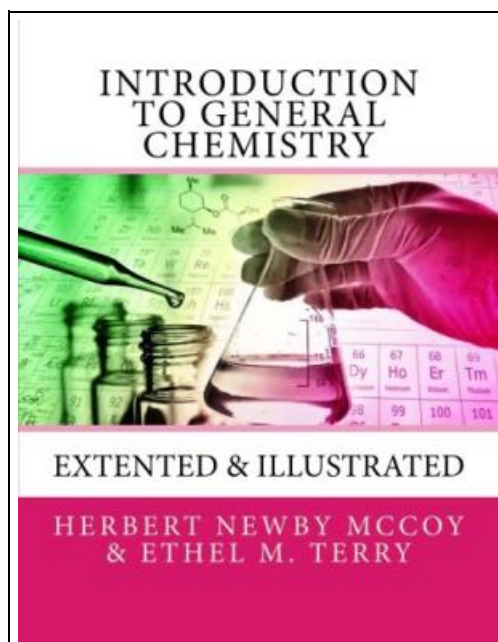


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Reviews

Very beneficial to all category of folks. We have study and that i am sure that i will planning to go through yet again again in the future. Its been printed in an extremely straightforward way in fact it is just soon after i finished reading this pdf where actually changed me, alter the way i really believe.

(Emmett Mann)

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Createspace, United States, 2015. Paperback. Book Condition: New. Expanded, Illustrated. 279 x 216 mm. Language: English . Brand New Book ***** Print on Demand *****.This text has been in process of growth at the University of Chicago since 1913. At that time a synopsis of the first nine chapters was printed. This was followed, in 1916-17, by the completed work of the first fifteen chapters. Chapters XVI-XIX were added and used in class work during 1918 and 1919. The book has been written for college Freshmen, and, as its title implies, it is intended to serve as an introduction to general chemistry. In consequence we have aimed to present a continuous and connected story in teachable form and have not attempted to give extensive descriptive and numerical data where such matter is of little interest to the student or is not needed for the development of important principles. Inasmuch as the choice and arrangement of topics in the earlier part of this book depart noticeably from the familiar order, some explanation seems necessary. We shall therefore sketch briefly the plan of the less conventional chapters, together with the philosophy of arrangement which has resulted in the scheme presented. The first chapter, which is brief, deals with the measurement of gases and the gas laws. In the next five chapters the most fundamental concepts of the science of chemistry are developed. These include: indestructibility of matter, idea of a pure substance, decomposition of pure substances, elements, analysis of substances and percentage composition, the law of definite composition, derivation of formulae. Chapter V shows how chemical formulae are derived from a knowledge of percentage composition and gas or vapor density. This development keeps as close as possible to the arguments of Avogadro and Cannizzaro and shows how formulae are obtained by methods...



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