# Researches on the Affinities of the Elements and on the Causes of the Chemical Similarity or Dissimilarity of Elements and Compounds



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(Lawrence Keeling)

## RESEARCHES ON THE AFFINITIES OF THE ELEMENTS AND ON THE CAUSES OF THE CHEMICAL SIMILARITY OR DISSIMILARITY OF ELEMENTS AND COMPOUNDS



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*\*. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1905 Excerpt: .as we pass towards A. H 1; Li 5; Na 5; Cu 3; Ag 0.5; Au 0.5. B, Al, Ga, In, Tl. So far as can be judged, the affinity seems to increase as we pass from B to Al, and then probably decreases until Tl is reached. B 3; Al 3; Ga 3; In 2; Tl 0.5. C, Si, Ge, Sn, Pb. C and Si have a very feeble affinity for P, Sn has a greater affinity. Pb also seems to have but a feeble affinity. C 1; Si 1; G 2; Sn 2; Pb 0.5. N, P, As, Sb, Bi. N has a very considerable affinity for N, and the affinity sinks from N to Bi, being very small at the latter point. N 12; P 7; As 1; Sb 0.5; Bi 0.5. O, S, Se, Te. O has a great affinity for P, and the affinity sinks steadily until Te is reached. O 9.25; S 4.0; Se 3.5; Te 2.0. F, Cl, Br I. The affinity for P for all these elements is great sinking from F to I. F 8.5; Cl 6.25; Br 4.15; I 2.5. Sulphur. Longitudinal Sections. Li, Be, B, C, N, P, F. The affinity of Li for S is considerable, but the affinity steadily decreases as we pass towards N, being very small here; it then increases again until F is reached. H 0.6; Li 14.5; Be 7.5; B 3.45; C 0.5; N 0.2; O 9; F 9....

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